		STUDY MODULE D	ESCRIPTION FORM			
Name o	f the module/subject cialist English		Code 1010702211010911730			
Field of	study		Profile of study	Year /Semester		
Che	Chemical Technology		(general academic, practical)	1/1		
Elective	path/specialty)	Subject offered in:	Course (compulsory, elective)		
	Orga	anic Technology	Polish	obligatory		
Cycle o	f study:		Form of study (full-time,part-time)			
Second-cycle studies			full-time			
No. of h	ours			No. of credits		
Lectu	re: - Classes	: 60 Laboratory: -	Project/seminars:	- 2		
Status o	of the course in the study	program (Basic, major, other)	(university-wide, from another fi	ield)		
		other	university-wide			
Educati	on areas and fields of sci	ence and art		ECTS distribution (number and %)		
techr	nical sciences			2 100%		
	Technical scie	ences		2 100%		
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Prere	quisites in term	s of knowledge, skills an	d social competencies:			
1	Knowledge	The already acquired language competence compatible with level B2 (CEFR)				
2	Skills	The ability to use general and field specific vocabulary, and grammatical structures required on the first level of studies				
3	Social competencies	The ability to work individually and in a group; the ability to use various sources of information and reference works.				
Assumptions and objectives of the course:						
 Advancing students? language competence towards the level at least B2+ (CEFR). Development of the ability to use academic and field specific language effectively in both receptive and productive language skills. 						
3. Imp	roving the ability to uno	derstand field specific texts (famili	arizing students with basic trans	slation techniques).		
4. Imp	roving the ability to fun	ction effectively on an internation	al market and on a daily basis.	a field of study		
Know		mes and reference to the	Euronanonai results for	a new or sludy		
NIIOV	neuye.					
SP:II-	•					
1. Give	•• e a talk on field specific riate linguistic and gra	or popular science topic (in Engl mmatical repertoire - [K_U01, K_	ish), and discuss general and fi _U04, K_U06]	eld specific issues using an		
2. Exp	ress basic mathematic	al formulas and to interpret data p	presented on graphs/diagrams	- [K_U02, K_U06]		
3. Conduct business correspondence in English - [K_U03, K_U04]						
4. Und 5. The English	erstand and analyze ir student is able to com	nternational, field specific literature municate effectively in a field spe	e - [K_U02, K_U03] ccific/professional area, and to g	ive a successful presentation in		
Social competencies:						
1. The student is able to recognize and understand cultural differences in a professional and private conversation, and in a different cultural environment [K K05]						
2. Stuc	2. Student understand the need for further education and improving the personal competence - [K_K01]					

Assessment methods of study outcomes					
-Formative assessment: tests during academic year (written and oral), presentations					
-Summative assessment: credit					
Course description					
1.Comparison of properties of glass and plastics					
2.Metals ? metallic bond, properties of metals, alloys, ways of hardening metals					
3. Colloids ? definition, phases, examples, emulsifiers, suspensions, solutions					
4. Cells, batteries, kinds of electrodes and electrolytes, rechargeable batteries, fuel cells					
5. Conductors and insulators ? testing electrical conductivity of different substances					
6.Properties of solids - the structure of diamond and graphite, allotropic forms					
7.Measurement of heat energy (combustion of ethanol and butane)					
8. Properties of ethanol, comparison of a structure of methanol, propanol, butane, ways of manufacturing ethanol ? comparison of the methods					
9. Kinds of radiation, radioactive decay of elements, making use of radioactivity					
Basic bibliography:					
1. Rose Marie Gallagher, Paul Ingram Complete Chemistry Oxford University Press 2000					
2. Maria Charmas English for Students of Chemistry M. C.Skłodowska University Press Lublin 2008					
Additional bibliography:					
1. Monika Korpak ??From Alchemy to Nanotechnology? Politechnika Krakowska 2008					
2. Piotr Domański English in Science and Technology Wydawnictwa Naukowo-Techniczne Warszawa 1993					
Result of average student's workload					
Activity	Time (working hours)				
1. Udział w ćwiczeniach		60			
2. Konsultacje do ćwiczeń	15				
3. Przygotowanie do ćwiczeń	45				
Student's workload					
Source of workload	hours	ECTS			
Total workload	120	2			
Contact hours	75	0			
Practical activities	60	0			